Cr(VI) is the chromium element in its 6+ oxidation state. Hexavalent Chromium can occur naturally in the environment from the erosion of natural chromium deposits or produced during industrial processes. Chromium VI can naturally occur from the transformation of Cr(III) under the right conditions such as exposure to UV light, high temperatures, low humidity, low pH-Values, or oxidizing agents. Chromium, in this oxidation state, can be extremely stable and persistent in the aquatic environment.

### Uses in the Supply Chain
Within the apparel and footwear supply chain, chromium may be in use in the following applications:

**Leather tanning:** Leather tanning is often performed using trivalent chromium (Cr(III)). Under certain conditions such as exposure to UV light, high temperatures, and/or low humidity, Cr(III) may transform into Cr(VI).

**Pigments and Dyes:** Cr(VI) may be used as a mordant after dyeing, especially for wool, or in lead-chromate pigments, inks, plastics, and surface coatings.

### Why Cr(VI) is Restricted
- Exposure to hexavalent chromium has been linked to many health problems in humans. Acute and long-term exposure to Cr(VI) has been linked to allergic skin reactions, gastrointestinal and respiratory issues, and damage to the male reproductive system. Cr(VI) is considered a carcinogen by the International Agency for Research on Cancer (IARC).¹
- Cr(VI) is regulated in textiles and leather products in multiple regions globally, including Egypt, Europe, South Korea, Taiwan and China.²,³

### Sourcing Compliant Materials from Your Suppliers
Contact your suppliers and explain that you require materials with extractable Cr(VI) values that do not exceed the AFIRM RSL listed limits.

In chromium tanned leather where Cr(III) is added to the hides, Cr(VI) may occur under certain circumstances. To help mitigate the risk of Cr(VI) generation, see the following guidance:

- **Use of chromium-based colorants and/or chromium-based after-treatments may cause these materials to fail the AFIRM limits. Compliance of chromium-based colorants available on the market has to be checked with the suppliers as described in the sections below.**
- **Share this guidance sheet with your material suppliers and instruct them to work with their chemical suppliers to source Cr(VI) compliant chemical formulations using the “Sourcing Chromium VI-Compliant Chemical...**

### May Be Found In
- Tanned natural leather
- Pigments and dyes (especially those used in wool)

### CAS Number

<table>
<thead>
<tr>
<th>Substance</th>
<th>CAS Number</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chromium VI (the element Cr in the 6+ oxidation state)</td>
<td>18540-29-9</td>
</tr>
</tbody>
</table>

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¹ International Agency for Research on Cancer (IARC).
² Regulations in Egypt.
³ Regulations in Europe, South Korea, Taiwan and China.
Preventing Cr (VI) Formation through Leather Tanning

- Roughly 80% of global leather production is chrome tanned. Alternatives are generally more expensive and produce different leather characteristics. Viable alternatives include:
  - Metal salts of aluminum, zirconium or titanium.
  - Synthetic organic compounds such as glutaraldehyde, tetrakis hydroxymethyl phosphonium sulfate (THPS), polymeric syntans or poly(carbamoyl)sulfonate (PCS).
- When tanning with Cr(III), the formation of Cr(VI) can be reduced by following these steps:
  - Utilize Cr(VI) free tanning products, tested for Cr(VI) and procured from a reputable supplier.
  - Avoid use of highly unsaturated fats:
    - If the raw material has a high level of natural grease, degrease thoroughly.
    - Avoid fat liquors or finishing oils with natural fats (e.g., fish, animal or vegetable).
  - Maintain pH below 6.5 and avoid use of ammonia, sodium bicarbonate and bleach during and after the tanning step.
  - Avoid the accumulation of free chromium on leather surface by washing after neutralization step, after any wet-end waterproofing treatment and at the end of wet processing.
  - Add antioxidant chemicals in re-tanning process (e.g., vegetable tannins).
  - Finish the wet end processing at low pH conditions (3.5 - 4)
- The transformation of the residual Cr(III) into Cr(VI) should be prevented also for the finished goods. For this reason, it is recommended to:
  - Avoid storage and transport of leathers in air polluted heavily with oxidizing gases like ozone, nitrogen oxides, sulfur oxides.
  - During transport, manufacture and use of leather items, avoid the prolonged exposure to low humidity, high temperatures and direct sunlight.
  - Avoid heating chrome-tanned leather to apply prints or set adhesives.

Sourcing Compliant Formulations from Your Chemical Suppliers

- Contact your chemical suppliers and explain that you require chemical formulations with no intentionally-added Cr(VI). The concentration of Cr(VI) should be <10 ppm (0.001%) in accordance with the ZDHC’s MRSL limits.  
  - Pay special attention to suppliers of wool after-dyeing mordants, metal-complex dyes and pigments and chromium leather-tanning powders.
- Have your chemical suppliers verify that their chemical formulations meet the <10 ppm Cr(VI) formulation limit with a certification or, if necessary, by providing a test report from a third party testing laboratory.
- Check the Safety Data Sheets (SDS) of all chemical formulations to ensure that Cr(VI) and chromate salts,
including dyes and pigments, are not listed as ingredients.

- Perform risk-based checks of your chemical suppliers’ formulations by submitting samples to a third party laboratory for testing to ensure the Cr(VI) <10 ppm limit is not exceeded.
- Discuss with your chemical supplier whether the below safer alternatives are suitable substitutes for your production needs.

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**Safer Alternatives**

- As mentioned previously, there are alternatives to the chrome tanning process, but these may result in leather with distinctly different performance properties.
- There are many alternatives to chromate pigments and dyes. Most major chemical suppliers have discontinued production of these dyestuffs.
- There are many after-dyeing mordant alternatives to chromate, including potassium aluminium sulfate, aluminium acetate and ferrous sulfate. Work with your chemical and dye suppliers to confirm that any chosen alternative is ZDHC MRSL compliant.

**Additional Information/ References**


